

Isolation of Phenolic compounds from ethyl acetate soluble fraction of the *Piper betle*

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Abstract—Four phenolics including pinoresinol-4-O- β -D-glucopyranoside (1), tachioside (2), 3,5-dimethoxy-4-hydroxyphenyl-O- β -D-glucopyranoside (3) and 3,4-dihydroxyallylbenzene (4) were isolated from the ethyl acetate soluble fraction of *Piper betle*. Their chemical structures were determined from ESI-MS, 1D-NMR, 2D-NMR spectral data and comparison to previous literature. With the exception of compound 4, this is the first report on the identification of compounds 1-3 from *P. betle*.

Keywords—*Piper betle*; phenolic; lignan; allylbenzene derivative;

I. INTRODUCTION

Piper betle L., a species of Piperaceae family, is a well-known herb and medicinal plant distributed widespread in tropical countries in the southern and southeastern Asia such as India, Sri Lanka, Malaysia, Thailand, Vietnam [1]. The leaves and roots of *P. betle* have a long history of use in traditional medicine for treatment of infectious diseases, sores, inflammation, flatulence, and cold [1, 2]. In term of the chemical constituents, several terpenoids, phenolics, alkaloids, steroids and aliphatic acids have been isolated from various parts of *P. betle* [3-5]. Its methanolic extract, essential oil, and some isolated compounds were investigated and showed a lot of valuable bioactivities such as anti-microbial activity, antiseptic, antidote, anti-inflammation, and anti-oxidant. In the aim to clarify chemical components of *P. betle* and support for explanation of its biological effects, herein, we report the isolation and structural elucidation of four phenolic compounds from ethyl acetate soluble fraction of the whole plants of *P. betle*

II. MATERIAL AND METHODS

A. General experimental procedures

NMR spectra were recorded on Jeol FT-NMR spectrometer (400 and 600 MHz). Column chromatography was performed using a silica gel (Kieselgel 60, 70 - 230 mesh and 230 - 400 mesh, Merck) or reversed phase C18 resins (150 μ m, YMC). Thin layer chromatography (TLC) was performed using a pre-coated silica gel 60 F₂₅₄ and/or RP-18 F_{254s} plates. The compounds were visualized by UV radiation (254 and 365 nm) and spraying with aqueous solution of sulfuric acid (C = 5%) followed by heating

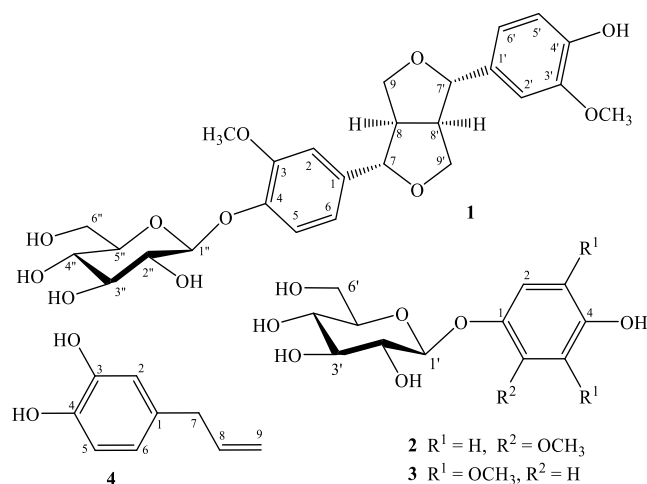


Fig. 1. Chemical structure of compounds 1-4

with a hot plate.

B. Extraction and Isolation

The dried and powdered whole plants of *P. betle* (3 kg) was macerated in MeOH for three times (each 6 L, 4 hrs in ultrasonic bath). Filtration and evaporation of the solvent under reduced pressure gave methanolic residue (170 g). This crude extract was then suspended in water (2 L) and successively partitioned with n-hexane, dichloromethane, and ethyl acetate. The ethyl acetate soluble extract (18 g) was fractionated on a silica gel column chromatography (CC) eluting with gradient solvent systems of CH₂Cl₂/MeOH (0-100% MeOH) to obtain five fractions (E1A - E1E). Fraction E1A was then purified by silica gel CC eluting with solvent system of EtOAc:CH₂Cl₂ (2/1, v/v) to obtain four smaller fractions (E1A1 - E1A4). Fraction E1A2 was continuously purified by silica gel CC using CH₂Cl₂/MeOH (7/1, v/v) to yield compound 4 (107 mg). Fraction E1D was purified by CC using solvent system of CH₂Cl₂/MeOH (5/1, v/v) to get four smaller fractions (E1D1 - E1D4). Fraction E1D1 was further purified by reversed phase C18 column chromatography with MeOH/H₂O (1/1, v/v) as eluent to afford compounds 2 (14 mg) and 3 (25 mg). Compound 1 (19 mg) was obtained from the fraction E1D3 on a reversed phase C18 column chromatography eluting with MeOH/H₂O (2/3, v/v).

- Pinoresinol-4-O- β -D-glucopyranoside (1): Pale yellow amorphous powder; ESI-MS *m/z*: 543 [M+Na]⁺;

$^1\text{H-NMR}$ (600 MHz, CD_3OD) and $^{13}\text{C-NMR}$ (150 MHz, CD_3OD) are given in the Table 1.

- Tachioside (**2**): White amorphous powder; ESI-MS m/z : 325 $[\text{M}+\text{Na}]^+$; $^1\text{H-NMR}$ (600 MHz, DMSO-d_6) and $^{13}\text{C-NMR}$ (150 MHz, DMSO-d_6) are given in the Table 2.

- 3,5-dimethoxy-4-hydroxyphenyl-*O*- β -D-glucopyranoside (**3**): White amorphous powder; ESI-MS m/z : 355 $[\text{M}+\text{Na}]^+$; $^1\text{H-NMR}$ (600 MHz, DMSO-d_6) and $^{13}\text{C-NMR}$ (150 MHz, DMSO-d_6) are given in the Table 2.

- 3,4-dihydroxyallylbenzene (**4**): Pale yellow amorphous powder; ESI-MS m/z : 151 $[\text{M}+\text{H}]^+$; $^1\text{H-NMR}$ (400 MHz, CDCl_3) δ_{H} ppm: 3.26 (1H, d, $J = 6.8$ Hz, H-7), 5.09 (2H, m, H-9), 5.93 (1H, m, H-8), 6.65 (1H, dd, $J = 2.0, 8.0$ Hz, H-6), 6.73 (1H, d, $J = 2.0$ Hz, H-2), 6.81 (1H, d, $J = 8.0$ Hz, H-5); $^{13}\text{C-NMR}$ (100 MHz, CDCl_3) δ_{C} ppm: 39.3 (C-7), 115.6 (C-9), 115.7 (C-2), 116.0 (C-5), 121.3 (C-6), 133.5 (C-1), 137.5 (C-8), 141.4 (C-3), 143.2 (C-4).

III. RESULTS AND DISCUSSION

Compound **1** was obtained as pale-yellow amorphous powder. The $^1\text{H-NMR}$ spectrum of **1** exhibited six signals of protons at aromatic area which belongs to two ABX spin coupled systems at δ_{H} 6.98 (1H, d, $J = 1.2$ Hz), 7.11 (1H, d, $J = 8.4$ Hz), 6.76 (1H, dd, $J = 1.2, 8.4$ Hz) and δ_{H} 6.91 (1H, d, $J = 1.2$ Hz), 6.74 (1H, d, $J = 8.4$ Hz), 6.88 (1H, dd, $J = 1.2, 8.4$ Hz); two oxygenated methines at δ_{H} 4.70 (1H, d, $J = 4.2$ Hz) và 4.67 (1H, d, $J = 4.2$ Hz); anomeric proton at δ_{H} 4.86 (1H, d, $J = 7.8$ Hz); and two singlet signals of methoxy groups at δ_{H} 3.81 và 3.82 (each 3H, s). The $^{13}\text{C-NMR}$ spectrum revealed 26 carbon signals. Among them, the signals at δ_{C} 102.7, 74.8, 77.8, 71.3, 78.1, 62.5 were characterized for a glucopyranosyl group. At the same time, the spin-spin coupling constant of anomeric proton $J = 7.8$ Hz indicated the β -glucopyranosyl linkage. Two carbon signals at δ_{C} 56.7, 56.4 were assigned to methoxy groups. Other carbon signals comprising twelve aromatic carbons in region δ_{C} 110.9 - 150.9 and six others at δ_{C} 87.4, 87.0, 72.7, 72.6, 55.5, 55.3 suggested for a lignan compounds. The chemical structure and NMR data assignment of compound **1** were further clarified by analyses of 2D - NMR experiments (Figure 1 and Table 1). Diagnostic HMBC correlations (Figure 2) were observed between H-2 (δ_{H} 6.98), H-6 (δ_{H} 6.76) with C-4 (δ_{C} 149.0); anomeric proton H-1" (δ_{H} 4.87) and C-4 (δ_{C} 149.0) indicating the *O*- β -glucopyranosyl group at C-4. The location of methoxy group at C-3 were clarified by HMBC correlations between H-5 (δ_{H} 7.11) with C-1 (δ_{C} 137.4)/ C-3 (δ_{C} 147.2), and methoxy proton (δ_{H} 3.82) with C-3 (δ_{C} 147.2). HMBC correlations of H-7 (δ_{H} 4.70) and C-1 (δ_{C} 137.4)/ C-2 (δ_{C} 111.5)/ C-6 (δ_{C} 119.7) were confirmed the linkage of benzene ring at C-7. Alternatively, in the HMBC spectrum, it can be observed the correlation of H-7 (δ_{H} 4.70) and C-9' (δ_{C} 72.6), which was proved the formation of ether-cyclic between C-7 and C-9'. Similarly, the HMBC correlations of H-2' (δ_{H} 6.91), H-6' (δ_{H} 6.88) to C-4' (δ_{C}

150.9) also allowed to assign NMR data of C-2', C-4', and C-6'. The HMBC correlations of H-5' (δ_{H} 6.74) to C-3' (δ_{C} 147.4), C-1' (δ_{C} 133.7) and methoxy protons (δ_{H} 3.81) to C-3' (δ_{C} 147.4) suggested for the position of other methoxy group at C-3'. Another benzene ring was linked at C-7' confirming by HMBC correlations between H-7' (δ_{H} 4.67) and C-1' (δ_{C} 133.7), C-2' (δ_{C} 110.9), C-6' (δ_{C} 120.0). Also, the HMBC correlation of H-7' (δ_{H} 4.67) to C-9 (δ_{C} 72.7) allowed to determine ether linkage between C-7' and C-9'. Therefore, chemical structure of compound **1** was determined to be pinoresinol-4-*O*- β -D-glucopyranoside. The $^{13}\text{C-NMR}$ spectrum of compound **1** was also similar to that of pinoresinol-4-*O*- β -D-glucopyranoside in the literature [6]. The ESI-MS of **1** revealed a quasi-molecular ion peak at m/z 543 which was well agreed with its molecular formula of $\text{C}_{26}\text{H}_{32}\text{O}_{11}$. Recently, this compound was found in some plants such as *Forsythia suspense* [7], *Lippia triphylla* [8], *Nepenthes mirabilis* [9]. It was reported exhibiting anti-inflammatory effect, anti-oxidation, and protection of nerve cells. However, until now, there have not been any reports on the identification of this compound from *P. betle*

Compound **2** was isolated as a white amorphous powder. The $^1\text{H-NMR}$ analysis of **2** showed three aromatic protons with ABX spin-spin coupled system [δ_{H} 6.68 (1H, d, $J = 2.4$ Hz), 6.45 (1H, dd, $J = 2.4, 8.4$ Hz), and 6.64 (1H, d, $J = 8.4$ Hz)], an anomeric proton

TABLE I. $^1\text{H-}$ AND $^{13}\text{C-NMR}$ SPECTRAL DATA OF COMPOUND 1

No.	$\delta_{\text{C}}^{\text{a,b}}$	$\delta_{\text{H}}^{\text{a,c}}$ (mult., J in Hz)
1	137.4	-
2	111.5	6.98 (d, 1.2)
3	147.2	-
4	149.0	-
5	117.9	7.11 (d, 8.4)
6	119.7	6.76 (dd, 1.2, 8.4)
7	87.0	4.70 (d, 4.2)
8	55.3	3.07 (m)
9	72.7	4.19 (m)/ 3.82 (m)
1'	133.7	-
2'	110.9	6.91 (d, 1.2)
3'	147.4	-
4'	150.9	-
5'	116.1	6.74 (d, 8.4)
6'	120.0	6.88 (dd, 1.2, 8.4)
7'	87.4	4.67 (d, 4.2)
8'	55.5	3.07 (m)
9'	72.6	4.19 (m)/ 3.82 (m)
OGlc		
1"	102.7	4.87 (d, 7.8)
2"	74.8	3.48 (dd, 7.8, 9.6)
3"	77.8	3.45 (dd, 9.9, 9.6)
4"	71.3	3.37 (m)
5"	78.1	3.37 (m)
6"	62.5	3.84 (br d, 12.0)
		3.66 (dd, 4.8, 12.0)
3-OCH3	56.7	3.82 (s)
3'-OCH3	56.4	3.81 (s)

Measured in $^{\text{a}}\text{CD}_3\text{OD}$, $^{\text{b}}150\text{MHz}$, $^{\text{c}}600\text{MHz}$

TABLE II. ¹H- AND ¹³C-NMR SPECTRAL DATA OF COMPOUNDS 2 AND 3

No.	Comp. 2		Comp. 3	
	$\delta_C^{a,b}$	$\delta_H^{a,c}$ (mult., J in Hz)	$\delta_C^{a,b}$	$\delta_H^{a,c}$ (mult., J in Hz)
1	151.3	-	150.8	-
2	148.3	-	95.9	6.35 (s)
3	103.0	6.65 (d, 2.4)	148.7	-
4	141.8	-	130.8	-
5	108.4	6.42 (dd, 2.4; 8.4)	148.7	-
6	115.7	6.62 (d, 8.4)	95.9	6.35 (s)
OGlc				
1'	102.2	4.63 (d, 7.2)	102.2	4.65 (d, 7.6)
2'	73.8	3.15 (dd, 7.2, 9.0)	73.8	3.16 (dd, 7.6, 9.0)
3'	77.6	3.23 (t, 9.0)	77.7	3.19 (t, 9.0)
4'	70.5	3.08 (t, 9.0)	70.6	3.05 (t, 9.0)
5'	77.2	3.21 (m)	77.3	3.24 (m)
6'	61.4	3.67 (br d, 12.0)	61.4	3.65 (br d, 12.0)
		3.41 (dd, 4.2, 12.0)		3.38 (dd, 4.2, 12.0)
2-OCH ₃	56.0	3.70 (s)		
3,5-OCH ₃			56.3	3.68 (s)

Measured in ^a)DMSO-d₆, ^b)150MHz, ^c)600MHz

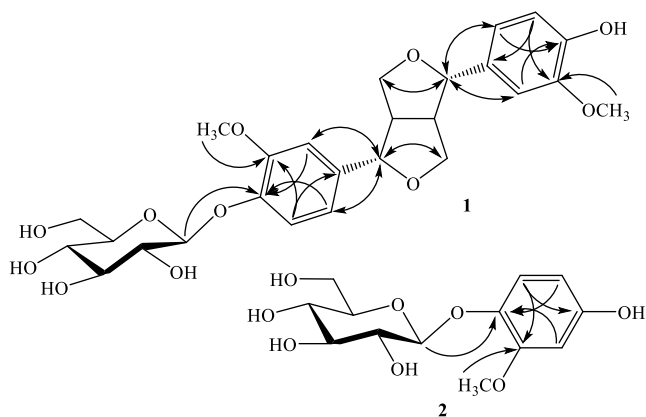


Fig. 2. Key HMBC correlations of representative compounds 1 and 2

[δ_H 4.66 (1H, d, $J = 7.2$ Hz)], and a methoxy group [δ_H 3.71 (3H, s)]. The ¹³C-NMR spectrum of 2 showed the signals of thirteen carbons. Similar to compound 1, six carbinol carbon signals (δ_C 102.2, 73.8, 77.6, 70.5, 77.2, 61.4) and J coupled constant of anomeric proton ($J = 7.2$ Hz) suggested the present of a β -glucopyranosyl group. The signal at δ_C 56.0 was assigned to a methoxy group. Six remaining carbons at aromatic carbon area [δ_C 151.3, 148.3, 141.8, 115.7, 108.4, 103.0] were indicated for the presence of a benzene ring. Therefore, compound 2 was a phenyl glucoside analog. Analyses of HMBC of 2 revealed the correlations between H-3 (δ_H 6.68)/ H-5 (δ_H 6.45) and C-1 (δ_C 151.3), anomeric proton H-1' (δ_H 4.66) and C-1 indicating O-glucopyranosyl group at C-1. On the other hand, HMBC correlations between H-6 (δ_H 6.64) and C-2 (δ_C 148.3)/ C-4 (δ_C 141.8), methoxy proton (δ_H 3.71) and C-2 (δ_C 148.3), and de-shielded signals of C-2/ C-4 led to the elucidation of a methoxy group at C-2 and hydroxy group at C-4. Consequently, compound 2 was determined to be 4-hydroxy-2-methoxyphenyl glucopyranoside. The ¹³C-NMR analyses was also agreed to the reported data in literature [10]. Common name of this compound was tachioside. It was found in

some plants such as *Saprosma merrillii* [11], *Polygonum capitatum* [12] and reported as the constituent exhibiting anti-inflammatory, anti-oxidation activities of them. However, this compound was reported for the first time from *P. betle*.

Compound 3 was obtained as a white amorphous powder. The ¹H- and ¹³C-NMR spectrum of 3 was similar with that of compound 2 except signals corresponding to benzene ring. The appearance of aromatic proton signals at δ_H 6.35 (2H, s) suggested the presence of 1,3,4,5-tetrasubstituted benzene ring. Moreover, signals of methoxy groups at δ_H 3.68 (6H, s) and δ_C 56.3 indicated for two methoxy group at C-3 and C-5, forming symmetric structure. HMBC correlation between aromatic proton H-1' (δ_H 4.65) and C-1 (δ_C 150.8), chemical shift of C-4 (δ_C 130.8) indicated a structure as 3,5-dimethoxy-4-hydroxyphenyl-O- β -D-glucopyranoside. The NMR data of compound 3 was found to match with those in the literature [13]. Similar with compounds 1 and 2, this is the first finding of compound 3 from *P. betle*.

Compound 4 was isolated as pale-yellow amorphous powder. The ¹H-NMR spectrum of compound 4 showed ABX coupled aromatic protons [δ_H 6.73 (1H, d, $J = 2.0$ Hz), 6.81 (1H, d, $J = 8.0$ Hz), 6.65 (1H, dd, $J = 2.0, 8.0$ Hz)], a vinyl group (CH₂=CH) group [δ_H 5.09 (2H) and 5.93 (1H)], and a methylene group [δ_H 3.26 (2H, d, $J = 6.8$ Hz)]. The ¹³C-NMR spectrum of compound 4 exhibited the signals of nine carbons and divided by DEPT as three non-protonated carbons (δ_C 133.5, 141.4, 143.2), four methine (δ_C 115.7, 116.0, 121.3, 137.5), and two methylene (δ_C 115.6, 39.3). Among them, two de-shielded signals (δ_C 141.4, 143.2) suggested that they linked to hydroxyl groups. From above evidences compound 4 was predicted to be 3,4-dihydroxyallylbenzene. This deduction was strongly confirmed by good consistence between NMR spectral data of compound 4 to that of 3,4-dihydroxyallylbenzene in the literature [14] and its ESI-MS data showing quasi-molecular ion peak at m/z

151 [M+H]⁺). This compound is a common constituent and reported as one of the main ingredients of *Piper* species [15]

In conclusion, four phenolic compounds including pinoresinol-4-O-β-D-glucopyranoside (**1**), tachioside (**2**), 3,5-dimethoxy-4-hydroxyphenyl-O-β-D-glucopyranoside (**3**), and 3,4-dihydroxyallylbenzene (**4**) were isolated from ethyl acetate soluble fraction of the whole plants of *P. betle*. Their structures were determined by ESI-MS, NMR spectroscopic methods and well consisted with those reported in the literature. To the best of our knowledge, this is the first report of compounds **1-3** from *P. betle*.

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