

Mineralogical, Physico-Chemical Characterization And Ceramic Properties Of Babouantou Clay Materials (Haut-Nkam, West Cameroon)

Wouatong Armand Sylvain Ludovic
Department of Earth Sciences
Faculty of Sciences
University Dschang
P.O. Box: 67 Dschang, Cameroon
aslwouat@yahoo.com

Bomeni Isaac Yannick
Department of Earth Sciences
Faculty of Sciences
University Dschang
P.O. Box: 67 Dschang, Cameroon
isaacbomeni@yahoo.fr

Tchounang Kouonang Serge
Local Materials Promotion Authority (MIPROMALO),
Ministry of Scientific Research and Innovation,
Yaoundé, P. O. Box 2396 Yaoundé, Cameroon
sergetcko@yahoo.fr

Kamgang Kabeyene Beyala Véronique
Higher Teacher Training College, University of
Yaoundé I, P.O Box 47 Yaoundé, Cameroon
gemkruy@yahoo.fr

Njopwouo Daniel
Department of Inorganic Chemistry, Faculty of
Sciences, University of Yaoundé 1. P.O. Box: 812
Yaoundé, Cameroon.
dnjop@yahoo.fr

Abstract—This work concerns the mineralogical, physico-chemical characterization and ceramic properties of the Babouantou clay materials (lower area of the Western flank of the Bana volcano-plutonic complex). Four samples of clayey materials were collected from wells in Nguieu and Batchieu localities precisely from the illuvial and eluvial horizons (BE1, BE2, BE3 and BE4). Mineralogical analysis showed that they are made of relictual minerals (quartz and feldspar), clays mineral (kaolinite), weathering minerals (illite), hydroxides (gibbsite) and oxyhydroxydes (goethite). The chemistry shows a silico-aluminous character with moderate iron (Fe_2O_3) content and small quantities of alkali (Na_2O , K_2O) and alkaline earth metals (CaO). It is poor in

organic matter (< 1%). The physical parameters display a continuous and wide spread granulometry with variable plasticity (5-13). The ceramic tests conducted on fired bricks at 900°, 1000°, 1100°C respectively, presented some standard physico-mechanical properties which are acceptable in the domain of traditional ceramics. Based on test performed on fired bricks, these materials can be classified as either malleable fusible clays (samples BE1, BE3, BE4) and malleable refractory clays (sample BE2). Therefore, these clay materials can be used in ceramics at 1000°C of firing after crushing and screening.

Keywords—*Babouantou, clay materials, mineralogical, physico-chemical analysis and ceramics properties*

I. INTRODUCTION

Clays have been used by humans since older-days to make fire or growth bricks, kitchen materials and to design products further according to their particular properties before and after firing [25,31,33]. Today, clays are the most important materials used by manufacturing and environmental industries. With regards to their wide industrial applications, they are gradually becoming an indispensable tool for development in developing as well developed countries.

They play a vital role in economic development [16,23,32,47]. The use of clay covers many applications such as: ceramics, paper, paint, [10,36,50] rubber and plastics, insecticides, making of additives for food [1,45], cosmetics, pharmaceuticals, drilling fluids, fertilizer carriers and geochemical barriers [6,21,40,46].

In ceramic industry, properties of immense importance include: mineralogical and chemical

composition [45], malleability, thermal behavior, color and strength after cooking [4,9].

In Cameroon, most ceramic products are imported [34,45]. This point can be explained by the lack of local industries or poor evaluation of potentially available clay materials. For some decades, a lot of research has been made on clay materials in Cameroon, particularly those containing kaolinite [7,29,34,37,38,45] and smectite [28,35].

These studies were concerned with the mineralogical, physico-chemical properties, mechanical strength after firing and clay materials activation in order to increase reliable data base for the starting of industrial exploitation.

This paper is aimed at characterizing the mineralogical, physico-chemical and ceramic properties of some clay materials from Babouantou (West-Cameroon) in order to increase the data base of industrial application of Cameroonian clays.

II. MATERIAL AND METHODS

A. Raw materials

The locality of Babouantou (Figure 1) is situated at the lower Western flank of the Bana volcano-plutonic complex, 15 km from Bafang.

The prevailing climate is equatorial and a surrounding pseudo-altitude tropical type, with two seasons: a long rainy season from March to October (eight months) and a short dry season that range from November to March (4 months). The average annual precipitation is about 1734 mm and the average temperature is 22°C [13].

The hydrographic network of the study area is dendritic in the South and dendritic to Sub-parallel in the North. The landscape presents isolated hills of different altitudes separated by low marshy areas which serve as potential sites for clay deposit.

The samples analyzed during this study were collected from three trial pits particularly in alluvial and illuvial horizon developed on porphyritic granite in Babouantou.

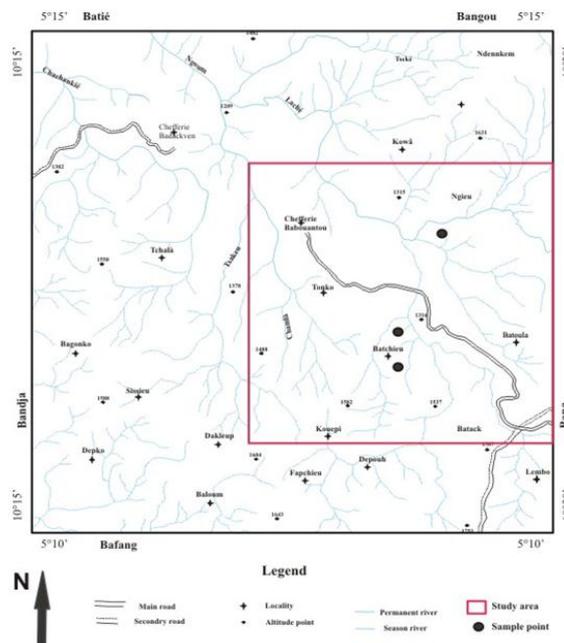


Figure. 1. Hydrographic map and sample point of clay materials

B. Mineralogical analysis

X-ray diffractometric method was used for mineralogical analysis in the mineralogy laboratory of the University of Liège (Belgium). Before analysis, the samples were first crushed in an agate mortar until the particle size is smaller than 50µm, then, dried at 105°C for 24 hours. Twelve milligrams of diffractogram

each sample were placed on the sample carrier of the diffractometer. The scanned angular range is between $5^\circ \leq 2\theta \leq 90^\circ$ with an angular pitch of 0.020° . The copper K α 1 radiation ($\lambda = 1.5406 \text{ \AA}$) used was produced at a voltage of 40 KV and a current 30 mA. The ASTM sheets are used to identify the minerals on.

C. Physico-chemical analysis

Particle size distribution and plasticity were considered as the two important parameters to evaluate the suitability of clays as raw material in ceramic domain [14,36]. Bulk chemical composition was also considered as a starting point for ceramic bodies [30].

The particle size was determined by wet sieving for fraction higher than eighty micron (80µm) and sedimentry for fraction less than eighty micron (80µm) according to [41].

Plasticity is a technological parameter that influences the characteristics of ceramic materials [19]. It was performed on fractions less than 400µm which consist of varying the water content on the material in order to evaluate its consistency.

The method called Atterberg's limits was used in order to define the consistence limit between the solid and the plastic state (plastic limit: W_P) and the plastic state to the liquid state (liquid limit: W_L). The interval between the plastic limit and liquidity defines the plasticity index (plasticity index: I_P).

The plasticity test was carried out according to the AFNOR prescription [42] in the material laboratory of MIPROMALO (Mission Local Materials Promotion) Yaoundé Cameroon.

Methylene blue test was used to determine the amount of clay content (quality and quantity) on clayey materials.

Chemical analysis was performed by X-ray fluorescence spectrometer. It is a quantitative analysis of the most stable oxides (major elements and minor elements) in the clay materials. The major elements were determined by assay ICP-AES (Inductively Coupled Plasma-Atomic Emission Spectrometry).

The proportions of major elements were calculated and expressed in the form of oxides percentage.

Organic matter was determined indirectly by oxidation with a potassium dichromate ($K_2Cr_2O_7$) solution in a highly acidic medium (H_2SO_4). The potassium dichromate solution is used in excess. This excess potassium dichromate is titrated with ferrous sulfate ($FeSO_4 \cdot 7H_2O$) to determine the quantity that is needed for the oxidation of organic carbon. The formula below is then used to compute the quantity of organic matter.

$$\%M.O = 1.724 \times \%C.O \quad (1)$$

M.O: organic matter;

C.O: organic carbon

D. Ceramic properties

The samples were homogenized and previously separated in four parts. Half a fraction is

removed and dried in an oven for 24 hours at a temperature of 105°C. The particle size was reduced by crushing in a porcelain mortar and sieved using a mesh of diameter 1 mm. The under flow obtained was humidified at 5%.

The test specimens (80mm× 40mm ×10 mm) were made using a 50 kN hydraulic press, and then dried at room temperature and in an oven at a temperature of 105°C for twenty four hours respectively.

The dried samples were fired at 900°C, 1000°C and 1100°C for a period of 5 hours at a heating rate of 5°C/min in an electric furnace-type FP 34 G in MIPROMALO ceramic laboratory.

The ceramic properties like colour, tone, firing shrinkage, compressive strength, flexural strength, water absorption were evaluated using [3] respectively

III. RESULTS AND DISCUSSION

A. Morpho- structural description of profiles

• Nguieu Profile

This profile (figure 2a) has as coordinates (North: 05°12'37, 9"; East 10°19'18, 5"; altitude: 1298 m). It is made of five horizons and the depth is about three hundred and seventy centimeters (370 cm)

Isalteritic horizon greater than 360 cm thick. It is reddish at polychrome trend. It has a clayey sand texture and a massive structure. The figurative elements are the ghosts' granite and fine quartz gravel. Its limit is distinct and progressive.

Alloteritic horizon has a thickness of about 25cm and is dark gray (5YR4 / 1: dark gray). Its texture is sandy silt and structure is lumpy. Its upper limit is frank and the lower limit is progressive.

Iluvial horizon (210 cm) is moonlight black (5YR4 / 2: dark gray). Its texture is sandy clay and the structure is massive. The limits are progressive. The upper limit is covered by a thick and bars of about 20 cm. This sandbar has a particular structure and as andy-loam texture.

eluvial horizon (60 cm) is yellowish to light red (7,5YR6 / 8:reddishyellow).

Its texture is sandy loam and the structure is lumpy. Its lower limit is brutal while the upper limit is diffuse

Mineral- organic horizon (50cm) is reddish brown (5YR4 / 4: reddish brown). Its texture is silty sand and the structure is lumpy. It contains many rootlets.

• **Batchieu Profile (Upper slope)**

This profile (figure 2b) has as coordinates (North:10°18'13,70;East:11'45,3'''';altitude:1312m). The depth is about 160 centimeters. It is constituted of four horizons.

Alloteritic horizon is constituted of a slightly thick alteration zone embedded in greyish matrix which characterizes the granite fragments in weathering process. The texture is sandy clay and the structure is massive. Its limit with the iluvial horizon is regular.

Iluvial horizon (62 cm) is light gray (5Y7 / 1: light gray). The texture is sandy clay and the structure is friable. Its boundaries are diffuse and regular. The sample (BE2) has been collected at this depth.

Eluvial horizon (64 cm) is pale brown (2,5Y8 / 2: pale brown). The texture is sandy clay and the structure is friable. On this horizon, we observe quartz gravel scattered throughout. Its upper limit is progressive.

Organo-mineral horizon is illustrated by the thick layer of about 24 centimeters and is yellowish red (5YR5 / 6: yellowish red). The texture is silty sand and the structure is friable. Here, we observe many roots with fine quartz gravel ranging from millimeter to centimeter

• **Batchieu Profile (middle slope)**

This mid-slope profile is described in a well of about 640 centimeters at coordinates points (North: 05 °11'45,4', East: 10°13'15,1'; altitude 1323m). It is constituted of six horizons (figure 2 c)

Isalteritic horizon has a variable coloration (reddish to yellowish). Its thickness is greater than 590 centimeters and the texture is sandy to sandy clay with a massive structure is massive. This horizon shows the ghosts of granite on weathering process. Its upper limit is diffuse and progressive.

Alloteritic horizon is about 80 centimeters thick. It has varying colours (brown, gray, and pale yellow). The texture is sandy clay and the structure is polyhedral. Its boundaries are irregular.

Iluvial horizon (250 cm) is reddish yellow (5YR6 / 8: reddish yellow). Its texture is sandy clay and the structure is lumpy. Its limits are regular. This horizon is crossed by a thick relictual vein of quartz-feldspar of about 14 centimeters. The sample (BE3) was taken from this horizon.

Primary eluvial horizon (125 cm) is yellow brown (10YR5 / 8: yellow brown). The texture is clay loam and the structure is polyhedral. The sample (BE4) was taken in this horizon.

Secondary eluvial horizon (120 cm) is yellowish red (5YR4 / 6: Yellow red). The texture is sandy clay and the structure is massive. Its upper and lower limits are progressive.

Organo-mineral horizon is represented by a thick layer of about 15 centimeters. The colour is dark red brown (2,5YR3 / 3: dark reddish brown). Its structure is lumpy and the texture is sandy loam. It contains rootlets. Its lower limit is diffuse.

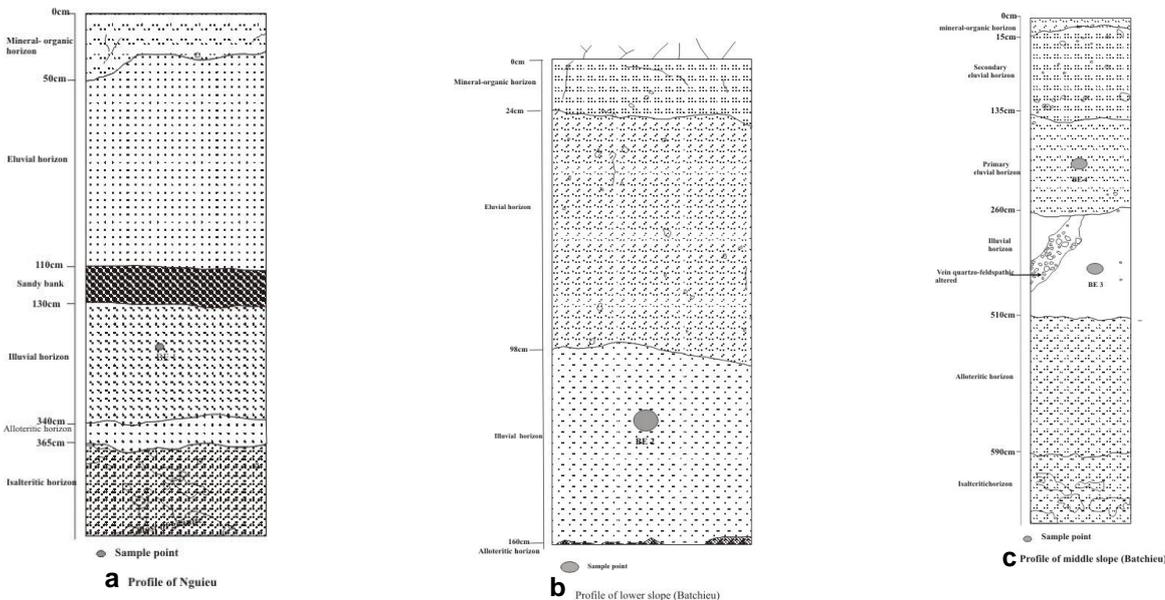


Figure.2. Profiles of different sample points

B. Description of sample collected in study area

TABLE I. CHARACTERISTIC OF SAMPLES COLLECTED FROM TRIAL PITS IN DIFFERENT LOCALITIES OF BABOUANTOU

Locality	Sample	Position	Depth (cm)	Thickness (cm)	Texture	Colour
Nguieu	BE1	Illuvial horizon	130	210	Clay sandy	Light black (5YR4/2)
Batchieu	BE2	Illuvial horizon	98	62	Clay sandy	greyish (5Y7/1)
	BE3	illuvial horizon	260	250	Sandy Clay	Reddishyellow 5YR6/8
	BE4	Elluvial horizon	135	125	Sandy Clay	Yellow brown 10YR5/8

C. Mineralogy

The main mineral phases (Figure 3) which are well expressed in these materials are: kaolinite (K), quartz (Q), illite (I), gibbsite (Gi), feldspar (Fd) and goethite (Goe). Table 2 below shows the relative proportions of minerals based on the peak intensity.

However, kaolinite, the most requested mineral in the ceramic industry [45] is abundant (32% - 41%), and feldspar and iron that carry ceramic properties. Thus, these materials can be used in ceramic paste.

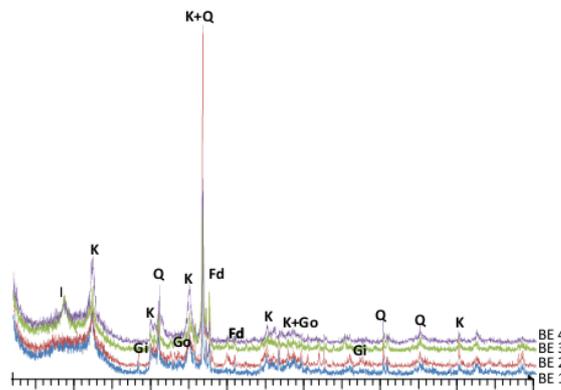


Figure 3. X-ray diffraction pattern (Cu $K\alpha_{1,2}$) of randomly non-oriented bulk sample showing the composition of sample analysis: I: illite; K: kaolinite; Gi: gibbsite; Go: goethite; Q: quartz; Fd: feldspath

TABLE II. MINERALOGICAL COMPOSITION OF SAMPLE STUDIED BASE ON SEMI QUANTITATIVE ANALYSIS (%)

Minerals	Kaolinite	Illite	Quartz	Feldspath (%)	Gibbsite	Goethite
BE1	35	18	24	16		7
BE2	32		26	27	8	7
BE3	33	31	8	21		7
BE4	41		20	10		29

D Physico-chemical analysis

The particle size distribution of clay materials (table 3) is homogeneous and wide spread where all fractions are represented. Sample BE2 is characterized by a high proportion of fine sand (48%). Sample BE1 has the highest rate of clay (28%) and sample BE3 has the highest percentage of silt (22%). Generally, the clay content of the material analyzed is between 16% and 28%. The

homogeneity and particle size distribution is consistent for the manufacture of firing bricks [22] and requires another processing such as crushing and screening before being used in pottery.

Atterberg's limit: table 3 shows that the samples BE1, BE2, BE3 and BE4 may contain 32%, 29%, 57% and 57% of water without sinking under their own weight. In addition, they can deform

plastically with lower water contents of 25%, 24%, 44% and 45% respectively.

The low plasticity (5-13) is due to the high content of sand [22]. Which is confirmed by the low absorption of methylene blue (table 3) by the samples.

According to the diagram, (figure 4), these samples are suitable for firing bricks.

TABLE III. PHYSICAL CHARACTERIZATION OF CLAYS SAMPLES COLLECTED ON BABOUANTOU LOCALITY (%)

Physic parameters	Samples types	BE1	BE2	BE3	BE4
Particule size	Gravel	0	6	3	6
	Sand	18	48	12	32
	Fine Sand	42	22	42	38
	Silts	14	8	22	8
	clay	28	16	21	1
Atterberg limit	W _l	32	29	57	57
	W _p	25	24	44	45
	I _p	7	5	13	12
Rate of methylen blue	VBS	1,13	0,93	2,80	1,33

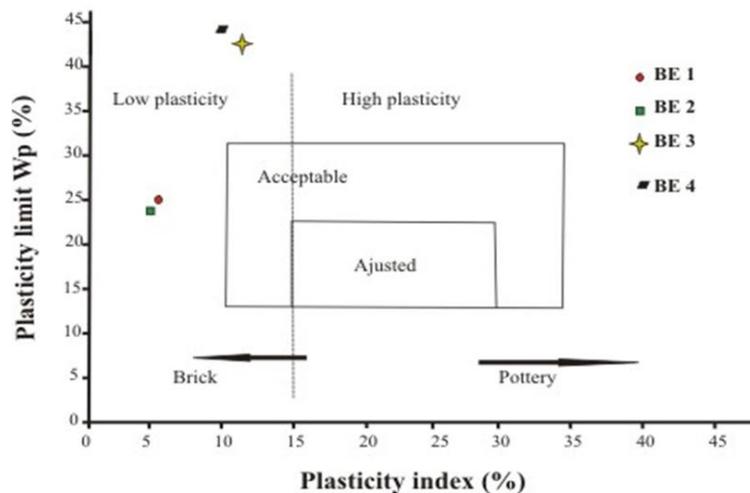


Figure. 4. Bain & Highley Diagram (1979) showing the position of our sample

- The geochemical analysis in table 4 shows a silico-aluminous character with moderate quantity of alkali and alkaline earth metals. In the words, the variation of potassium (K_2O : 2,74 -5,35) and iron (Fe_2O_3 : 1,94 – 8,39) confirms the presence of feldspar and illite [45] and an iron compound like hematite which is of big importance in the ceramic industry.

The SiO_2 / Al_2O_3 ratio is greater than 2% and the Fe_2O_3 / Al_2O_3 ratio lower than 1% respectively reflecting the abundance of quartz [37,45] and the presence of iron compound.

- Organic matter content is relatively low (<1%) and varies from 0.75% to 0.91%.

TABLE IV. MAJOR ELEMENT COMPOSITIONS (WT %) AND ORGANIC MATTER CONTENT IN CLAY MATERIALS FROM BABOUANTOU

Major elements (%)	BE1	BE2	BE3	BE4
SiO ₂	55,4	60,35	55,89	52,74
Al ₂ O ₃	23,13	21,83	20,38	23,27
Fe ₂ O ₃	4,13	1,94	6,98	8,39
MnO	0,02	0,01	0,02	0,01
MgO	1,04	0,51	1,65	0,75
CaO	0,32	0,18	0,21	0,02
NaO	0,10	0,14	/	/
K ₂ O	2,74	4,19	5,35	2,55
TiO ₂	1,52	2,14	1,43	0,99
P ₂ O ₅	0,11	0,09	0,22	0,19
LOI (1000°)	11,04	10,88	7,36	8,15
Total	99,99	100	99,99	99,99
SiO ₂ /Al ₂ O ₃	2,39	2,76	2,74	2,26
Fe ₂ O ₃ /Al ₂ O ₃	0,17	0,1	0,34	0,36
Organic matter	0,91	0,80	0,69	0,75

E Ceramic properties

- The colour of fired bricks (table 5) changes with an increase in temperature. This change in colour is related to the presence of goethite (FeOOH) or hematite [45] and the presence of Calcium (CaO), Magnesium (MgO), Titanium (TiO₂), Aluminum oxide (Al₂O₃) or temperature of the oven [5].

- Sound transmission is summarized on table 4. We noticed that, the specimens BE1, BE3 and BE4 have a metallic sound which is a characteristic of

good quality ceramic product [12]. However, the BE2 specimen doesn't have a metallic sound in the three temperature ranges. This lack of metallic sound is explained by the absence of conversion reaction during firing or the presence of refractory oxides such as silica (SiO₂), aluminum oxide (Al₂O₃), calcium oxide (CaO), magnesium oxide (MgO) which has not yet reached their respective melting temperatures (1710 °C, 2020 °C, 2570 °C and 2800°C).

TABLE V. PHYSICAL PROPERTIES OF CERAMIC TEST

Ceramic properties	Sample	900°C	1000°C	1100°C
Color	BE1	Yellowish light	Yellowish light	Yellowish light
	BE2	white Yellow	Whitishyellow	Whitishyellow
	BE3	Light red	Reddish	Red
	BE4	Light red	Reddish	Reddish
Sonority	BE1	Metallic sound	Metallic sound	Metallic sound
	BE2	Non metallic sound	Non metallic sound	Non metallic sound
	BE3	Metallic sound	Metallic sound	Metallic sound
	BE4	Metallic sound	Metallic sound	Metallic sound

- Firing shrinkage of the samples BE1 and BE4 (Figure 5) increases gradually from 1000 °C as a result of chemical transformation [17] marked by the passage from metakaolin to mullite [27] or the formation of a spinel phase or the appearance of alumina [45]. In addition, the linear shrinkage of the sample BE3 increases with temperature thus reflecting the vitrification initiated at temperatures below 900°C [49]. Furthermore, nonlinear shrinkage for the BE2 specimen confirms the absence of

chemical reaction during sintering or excess degreaser [26].

- Compressive strength is illustrated in figure 6. In this figure, we observed that the BE1, BE2 and BE3 specimen reach their maximum resistance at 1000°C with values between (10-20 MPa) generally recommended for traditional ceramics [12,20]. In contrary, the BE4 specimen begins its compressive strength from 1000° C. This behavior can be explained by their iron content (8.9% > 6% which is recommended for traditional ceramics [45] whose melting point is at 1200 °C where an eutectic

mixtures is formed [24]. This elevated temperature of iron confers poor textural properties to the sintered products

- Flexural strength of ceramic specimen (figure 7) increases gradually with temperature. This increase shows that, the vitreous phase in charge of the mechanical properties has not reached its optimum [11]. However, the flexural strength ranges from 0.36 to 4.32 MPa, but remains lower than

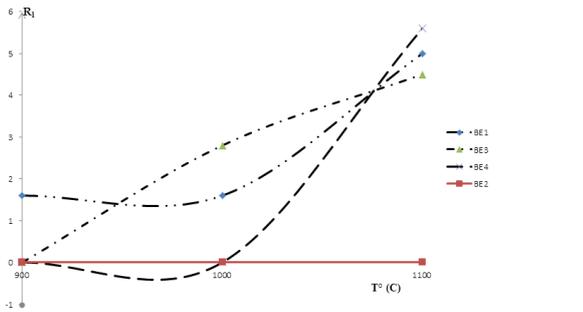


Figure.5. Variation of linear shrinkage with the temperature

5 MPa which corresponds to the lower limit for traditional ceramics [12].

- Water absorption (figure 8) decreases from 1000 °C following the appearance of the vitreous phase which fills the pores between the grains during firing [24]. Thus, the temperature of 1000 °C is recommended to reduce the porosity of the products during sintering.

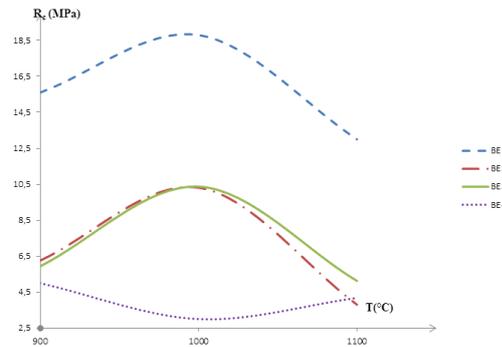


Figure. 6. Variation of compressive strength with the temperature

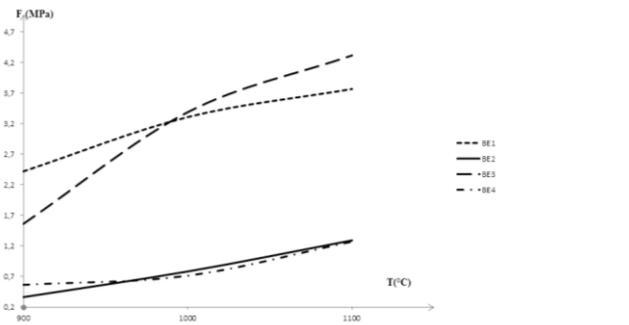


Figure. 7. Variation of flexural strength with the temperature

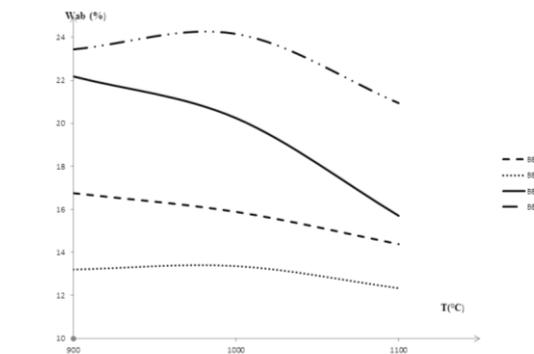


Figure. 8. Variation of water absorption with the temperature

IV CONCLUSION

The Babouantou clay materials are developed on porphyritic granite and concentrated principally on the iluvial and eluvial horizons. The mineralogical analysis of four samples (BE1, BE2, BE3 and BE4) is made up essentially of kaolinite, illite, quartz, orthoclase, goethite and gibbsite accessorially. Physical properties showed a homogeneous wide spread granulometry. It has a low plasticity (5-13) and confirms the low methylene blue absorption rate. Chemical analysis reveals the silico-aluminous character of all samples with varying amounts of iron (FeO) and smaller alkaline (Na₂O, K₂O) and alkaline earth metal (CaO).

The ceramic test on firing bricks reveals that the physical parameters (color, sound, linear shrinkage, absorption rate) and mechanical parameters (flexural strength and compressive strength) are acceptable in the standards of traditional ceramics. According to the plasticity and sintering temperature, the samples BE1, BE3, BE4

are a little bit plastics and fusible while sample BE2 is a bit plastic and refractory. These clay materials may be used for the manufacture of firing bricks and roofing tiles after crushing, sieve and sintering at 1000°C.

BIBLIOGRAPHY

- [1] Allo.W.A; Murray.H.H. (2004). Mineralogy, chemistry, and potential applications of bentonite in San Juan Province, Argentina. *Appl Clay Sci.* 25, 237-243
- [2] ASTM- America Society for Testing Materials (1972) "water Absorption, Bulk Density, Apparent porosity and Apparent specific gravity of fired white ware products" C 373-72
- [3] ASTM-America Society for testing and materials (1977), "Flexural properties of ceramic white ware materials" C 674-77

- [4] Baccour.H; Medhioub.M; Jamoussi.F; Mhiri.T (2009). Influence of firing temperature on the ceramic properties of Triassic clays from Tunisia. *J. Mater. Process. Technol* 2009, 2812-2817
- [5] Boch.P. (2001). Frittage et microstructure des céramiques et processus céramiques. ed par Boch. P paris: Hermès Sciences publication P 73-112
- [6] Bertolino.L.C; Rossi.A.M; Scorzelli.R.B; Torem.M.L. (2010). Influence of iron on kaolin whiteness. An electron paramagnetic resonance study. *Appl Clay Sci.* 49, 170-175
- [7] Bomeni.I.Y (2013). Caractérisation minéralogique et physico-chimique des matériaux argileux de Babouantou (Ouest Cameroun) : Valorisation dans la céramique traditionnelle, thèse de Master en Science de la terre, Univ Dschang, 91p
- [8] Borthakur.P.C. (2003). Characterization, beneficiation and utilization of kaolin clay from Assam, India. *Appl Clay Sci.* 24, 93-103
- [9] Burst. J.F (1991). The application of the clay minerals in ceramics appl. *Clay Sci.*5, 421-443
- [10] Celik.H. (2010). Technological characterization and industrial application of two Turkish clays, for ceramic industry. *Appl Clay Sci.* 50, 245-254
- [11] Chinje.Melo.UF; Kamseu.E; Djangang(2001). Effect of fluxes on the fired properties between 950-1050°C of some Cameroonian clays. Actes 1ère conférence GCA, p57-69
- [12] Damle.A (2008). Training on fire brick production and entreprise viability. Plots N°98, Lane 5, Natraj Society Karvenagar. Pune-411 052, India
- [13] Dongmo.J. (1981). Le dynamisme Bamiléké. La maîtrise de l'espace agraire. Vol. 1, CEPER, Yaoundé, 427 p
- [14] Dondi.M; Fabbri.B; Guarini.G (1998). Grain -size distribution of Italian raw materials for building clay products a reappraisal of the winkler diagram. *Clay Miner.*33, 435-442
- [15] Duplain.R ; Lanchon.R ; Saint Arroman.J.C (2000). Granulats, sols, ciments et bétons. Caractérisation des matériaux de génie civil par les essais de laboratoire. Collection A. Capliez, Nouvelle édition, 237p.
- [16] Ekosse. I.G (1994). Clays: a gateway into the future. *Botsw Notes. Rec.* 26, 139-149
- [17] Elimbi.A; Njopwouo.D; Pialy.P; Wandji.R (2001). Propriétés des produits de cuisson de deux argiles kaoliniques de l'ouest Cameroun, *Sil.Ind.*, 66 [11-12], 121-125.
- [18] Fan.H; Song.B; Li.Q (2006). Thermal behavior of goethite during transformation to hematite. *Materials chemistry and physics*, Vol 98, 148-153
- [19] Guerraoui.F ; Zamma.M ; Ibnoussina.M., (2008). Caractérisation minéralogique et géotechnique des argiles utilisées dans la céramique de Safi (Maroc). *African Journal of Science and Technology, Science and Engineering Series* (9) 1, 1-11
- [20] Karfa.T (2003). Frittage à basse température d'une argile kaolinique du Burkinafaso: transformation thermiques et réorganisation structurale. Univ. Limoges/Fac Sciences et techniques. Thèse de Doctorat (n°15-2003).190p+annexe.
- [21] Konta.J (1995). Clay and man: clay raw materials in the services of man. *Appl Clay Sci.* 10, 275-335.
- [22] Kouayep Lawou.S (2006). Caractérisation géotechnique et propriétés des briques de terre cuites des matériaux d'altération développés sur les projections pyroclastites de la région de Bana (Ouest-Cameroun). Mémoire de DEA. Univ. Yaoundé I.75p+annexe
- [23] Kühnel.R.A. (1990). Modern days of clays. *Appl Clay Sci.* 5, 135-143
- [24] Lambercy.E(1993). Les matières premières céramiques et leurs transformations par le feu. *Granit1.Des dossiers argile.*509p
- [25] Lee.V.G ; Yeh. T.H (2008). Sintering effects on the development of mechanical properties of fired clay ceramics. *Mat. Sci. eng. A* 485, 5-13
- [26] Lemougna Ninla.P (2004). Contribution à la diminution des casses dans la production des briques de terre cuites à la PROMOBAT. Mémoire de D.E.S.S. Univ de Yaoundé I 58p
- [27] Lemaitre.J ; Leonard.A.J ; Delmon.B (1977). The sequence of phases in the 900-1050°C transformation of metakaolinite, *Proc. Int. Clay Conf.*, 60[1-2]p37-43
- [28] Mache (2013). Minéralogie et propriétés physico-chimiques des smectites de Bana (ouest-Cameroun) et les smectites de Sabga (Nord-Ouest) : utilisation dans la décoloration d'une huile végétale alimentaire. Thèse de doctorat, Univ de Liège Belgique, Université de Yaoundé1-Cameroun 145 p
- [29] Mamba.Mpelle. (1997). Essais de formulation pour la fabrication des carreaux céramiques à partir d'une argile de Yaoundé. *Journ Cam.Buld Mat.*Vol.1N°1.16-20
- [30] Manning.D.A.C (1995). Introduction to industrial mineral. Chapman & Hall edition London; 275p
- [31] Mohmudi.S; Srasra.E; Zargouni.F (2008). The use of Tunisian Barremian clay in the traditional ceramic industry: optimization of ceramic properties. *Appl. Clay Sci.*45, 125-129
- [32] Murray.H.H (2007). Applied clay mineralogy. *Developments in Clay Science*2. Elsevier B.V

- [33] Murray.H.H (2000). Traditional and new applications for kaolin, smectite, and palygorskite: a general overview. *Appl Clay Sci.* 17, 207-221
- [34] Ngon Ngon.G.F; Lecomte Nana.G.L; Yongue Fouateu.R; Lecomte.G; Bilong.P (2013). Physicochemical and mechanical characterization of ceramic materials obtained from a mixture of silica, Feldspars and clays material of the Douala Region in Cameroon (central Africa) *Advances in Ceramic Science and Engineering* 31p Vol 2
- [35] Nguetnkam.J.P (2004). Les argiles des vertisols et des sols fersiallitiques de l'Extrême- Nord du Cameroun: genèse, propriétés cristalochimiques et texturales, typologie et applications à la décoloration des huiles végétales. Thèse de doctorat d'État de l'Univ de Yaoundé I,216p
- [36] Ngun.B.K; Mohamad.H; Sulaiman.S.K; Okada.K; Ahmad.Z.A (2011). Some ceramic properties of clays from central Cambodia. *Appl Clay Sci.*53, 33-41
- [37] Njoya.A (2006). Etude du gisement de Kaolin de Mayouom (Ouest Cameroun): Cartographie, Minéralogie et Géochimie. Thèse de doctorat Univ de Yaoundé I,142p
- [38] Njoya.A; Nkoumbou.C; Grosbois.C; Njopwouo.D; Njoya.D; Courtin-No made, A., Yvon.J; Martin.F (2006). Genesis of Mayouom kaolin deposit (West Cameroon). *Appl Clay Sci.* 32, 125-140.
- [39] Njopwouo.D(1984). Minéralogie et physico-chimie des argiles de Bomkoul et de Balengou (Cameroun): Utilisation dans la polymérisation du styrène et dans le renforcement du caoutchouc naturel. Thèse de Doctorat d'Etat, Univ. Yaoundé, Cameroun 300p
- [40] Nkoumbou.C; Njoya. A; Njopwouo.D; Wandji.R. (2001). Intérêt économique des matériaux argileux au Cameroun. *Proceedings of the first conference on the valorization of clay materials in Cameroon, and launching of the Cameroonian clay group.* Yaoundé, April 11-12, 2001
- [41] Norme NF P 94-057 (1992). Analyse granulométrique par sédimentométrie. *Afnor* 15p
- [42] Norme NF P94-051 (1993). Reconnaissance et essai de détermination des limites d'Atterberg. *AFNOR.* 15p
- [43] Ollivier.G ; Bihannic.I ; Njopwouo.D ; Yvon.J ; and Bonnet. J.P (2008). Characterization for industrial applications of clays from Lembo deposit, Mount Bana (Cameroon). *Clay minerals* 43,415-435
- [44] Pialy.P ; Nkoumbou.C; Villiéras. F; Razafiti anamaharav.A; Barres.O; Pelletier.M; Ollivier.G; Bihannic.I; Njopwouo.D; Yvon.J; and Bonnet. J.P (2008). Characterization for industrial applications of clays from Lembo deposit, Mount Bana (Cameroon). *Clay minerals* 43,415-435
- [45] Pialy.P (2009). Étude de quelques matériaux argileux du site de Lembo (Cameroun): minéralogie, comportement au frittage et analyse des propriétés d'élasticité. Thèse de doctorat. Univ. LIMOGES.130p
- [46] Saikia.N.J Bharali.D.J ; Sengupta. P ; Bordoloi.D ; Goswamee.R.L ; Saikia.P.C ; Ristic.M; Degrave.E; Music.S; Popovic.S; OREHOVEC.Z (2007). Transformation of low crystalline ferrihydrite to α -Fe₂O₃ in the solid state. *Journal of molecular structure*; Vol 834-836pp454-460.
- [47] Reaves. G.M; Sims.I; Cripps.J.C (2006). *Clay minerals used in construction.* Geological Society. London, Engineering Geology Special Publication, 21.
- [48] Wouatong.A.S.L (1997). Mineralogical study of weathering on the Bana complex, West part of Cameroon. *J.Sci.Hiroshima Univ.Ser.CII(1)*, pp 1-43
- [49] Yakoubi.N ; Hamed Aberkan.M ; Mohamed.Ouadia (2006). Potentialité d'utilisation d'argiles marocaines de JbelKharrou dans l'industrie céramique. *Science Direct C.R Geoscience* 693-702p
- [50] Zhang.D; Zhou.C.H; Lin.C.X; Tong.D.S; Yu.W.H (2010). Synthesis of clay minerals. *Appl Clay Sci.* 50, 1-11.